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level to a lower energy level as shown. photon electron

- $5.40 \times 10^{-19} \text{ J}$ - $21.8 \times 10^{-19} \text{ J}$ The wavelength of the radiation emitted due to an electron transition between the two energy levels shown is A $7.31 \times 10^{-8} \text{ m}$ B $\times 9.12 \times 10^{-8} \text{ m}$ C $71.21 \times 10^{-8} \text{ m}$ D $68.23 \times 10^{-8} \text{ m}$ E $152.47 \times 10^{-8} \text{ m}$. 19. A ray of red light travels from air into water.

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