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Professor Paul H. Emmett  
Chemistry Department  
Portland State University  
P.O. Box 751 Portland,

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Oregon 97207 June 17, 1977

the fact that nitrogen at  $-195^{\circ}\text{C}$  is about 5% imperfect at atmospheric pressure, the correction being a linear function of the pressure. Instead of a curve with kinks a smooth S-shaped isotherm was obtained with nitrogen at  $-195^{\circ}\text{C}$ .

~~Brunauer S, Emmett P H &  
Teller E. Adsorption of  
gases in ...~~

Paul Hugh Emmett was an American chemist best known for his pioneering work in the field of catalysis and for his work on the Manhattan Project during World War II. He spearheaded the research to separate

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isotopes of uranium and to develop a corrosive uranium gas. Emmett also made significant contributions to BET Theory which explains the relationship between surface area and gas adsorption. He served on the faculty of Johns Hopkins University for 23 years throughout his scientific career.

~~Paul Hugh Emmett — Wikipedia~~  
S. Brunauer, P. H. Emmett and E. Teller, "Adsorption of Gases in Multimolecular Layers," Journal of the American Chemical Society, Vol. 60, No. 2, 1938, pp. 309-319.

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~~S. Brunauer, P. H. Emmett  
and E. Teller, "Adsorption  
of ...~~

The gas adsorption technique is based on the Brunauer-Emmett-Teller (BET) model included in the determination of the amount of gas that is adsorbed on the surface of the sample at low temperatures [13]. The specific surface area ( $m^2/g$ ) defined by this method comprises the outward as well as the internal (open pores) surface area. However, the surface area of closed pores cannot be determined because of its inaccessibility for adsorbing gas molecules.



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~~Brunauer Emmett Teller Model  
— an overview |  
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E Adsorption Of Gases In~~  
By making use of the  
evaporation-condensation

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properties of liquid  
mixtures, the  
Brunauer-Emmett-Teller  
theory of multimolecular  
adsorption extended to  
mixtures of gases. No  
satisfactory experimental  
data are available at the  
present time with which to  
test the theory.

~~Theory of Multimolecular  
Adsorption from a Mixture of~~

...

Emmett PH, Brunauer S, Love  
KS. The measurement of  
surface areas of soils and  
soil colloids by the use of  
low temperature van der  
waals adsorption isotherms  
Soil Science. 45: 57-65. 1:  
1937: Emmett PH, Brunauer S.

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Accumulation of alkali  
promoters on surfaces of  
iron synthetic ammonia  
catalysts Journal of the  
American Chemical Society.  
59: 310-315. 1: 1937

~~Paul Hugh Emmett~~

~~Publications~~

310 STEPHEN BRUNAUER, P. H.  
EMMETT AND EDWARD TELLER

Vol. 60 DeB~er~'~ in several  
papers showed that various  
experimental adsorption  
isotherms could be fitted by  
equation (4a). However, they  
could not evaluate  $K_1$   
because in all cases, except  
one, the surface, and  
therefore  $v_m$ , was not known.

~~[CONTRIBUTION FROM THE~~

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~~BUREAU OF Adsorption Gases  
in ...~~

P. H. Emmett's 15 research works with 1,206 citations and 356 reads, including:  
THE APPLICATION OF POLANYI'S POTENTIAL THEORY TO THE VAN DER WAALS ADSORPTION OF GASES ON IRON SYNTHETIC AMMONIA ...

~~P. H. Emmett's research works~~

Brunauer-Emmett-Teller theory aims to explain the physical adsorption of gas molecules on a solid surface and serves as the basis for an important analysis technique for the measurement of the specific surface area of materials.

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The observations are very often referred to as physical adsorption or physisorption. In 1938, Stephen Brunauer, Paul Hugh Emmett, and Edward Teller published the first article about the BET theory in the Journal of the American Chemical Society. The BET theory ...

~~BET theory~~ — Wikipedia  
Stephen Brunauer, P. H. Emmett, and ; Edward Teller;  
Journal of the American Chemical Society 1938, ...  
Herbert S. Harned; Journal of the American Chemical Society 1938, ...

~~Journal of the American~~

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~~Chemical Society | Vol 60,  
No 2~~

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Martyn S. Brown, Silvia  
Favero, Martin P. Attfield,  
Flor Siperstein, David  
Fairen-Jimenez, Klaus  
Hellgardt, Russell Purves,  
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Screening Metal-Organic

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Frameworks for Dynamic CO/N<sub>2</sub>  
Separation Using  
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~~Adsorption of Gases in  
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Brunauer, S., Emmett, P.H.  
and Teller, E. (1938)  
Adsorption of Gases in  
Multimolecular Layers.  
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Chemical Society, 60,  
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~~Brunauer, S., Emmett, P.H.  
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Adsorption Of Gases  
Incoun), by latest (which



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~~Brunauer S Emmett P H Teller E Adsorption Of Gases In~~  
Brunauer, S., Emmett, P.H., Teller, E., "Adsorption of gases in multimolecular layers," Journal of the American Chemical Society, 60 (2). 309-319. 1938.

~~Brunauer, S., Emmett, P.H., Teller, E., "Adsorption of~~

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## ~~Adsorption~~ Adsorption Of Gases In

The Brunauer-Emmet-Teller (BET) equation (Brunauer et al., 1938) overcomes the limitation to a monolayer of sorbate and currently is a commonly used model to describe the multilayer adsorption of gas molecules on a sorbent (Brunauer et al., 1938). For water adsorption, the BET equation can be written as

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