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133 20. If a hydrogen atom and a helium atom are traveling at the same speed, A) the wavelength of the hydrogen atom will be about 4 times longer than the wavelength of the helium atom. B) the wavelength of the hydrogen atom will be about 2 times longer than the

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Chapter 7. Quantum Theory and Atomic Structure • A problem arose in Rutherford's nuclear model. A nucleus and electron attract each other; to remain apart the electron must move.

- The energy of the electron's movement must balance the energy of attraction.
- Physics established that a charged particle moving in a curved

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Hydrogen has a red emission line at 656.3 nm, what is the energy and frequency of a photon of this light?

Note: Planc k's constant =  $6.626 \times 10$

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~~A 34 J·s, speed of light =  $2.998 \times 10^8$  m/s~~  
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7-8 The Quantum Theory of Energy  
Any object (including atoms) can emit or absorb only certain quantities of energy. Energy is quantized; it occurs in fixed quantities, rather than being continuous. Each fixed quantity of energy is called a quantum. Energy can be gained or lost only in whole number multiples of. An atom changes

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