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the Periodic Table

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Covalent - Double  
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are called valence  
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electrons located farthest from the nucleus.

Valence electrons are always in the highest energy level.

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Hour Information:

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electrons. Valence

electrons are the

electrons located farthest

from the nucleus. Valence

electrons are always in

the highest energy level

The valence electrons are

the most important

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electrons in an atom  
because they are the ...

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Home

Information: Steps for  
Drawing Lewis Structures  
for Covalent

Compounds Study the  
two examples in the table  
of how to write structures  
for  $\text{CO}_3^{2-}$  and  $\text{NH}_3$ .

Make sure you  
understand each of the

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five steps.  $\text{CO}_3^{2-}\text{NH}_3$

Step #1: Add up the number of valence electrons that should be included in the Lewis Structure.  $4 + 3(6) + 2 = 24$  (carbon has four and each

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