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Combustion Processes $C + O_2 \rightleftharpoons$

CO_2 $C + 0.5O_2 \rightleftharpoons CO$ $2H + 0.5O_2$

$2 \rightleftharpoons H_2O$ $S + O_2 \rightleftharpoons SO_2$

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APPENDIX F HEATS OF FORMATION

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AND COMBUSTION TABLE F.1 Heats of Formation and Heats of Combustion of Compounds at 250C*t Standard states of products for are C02(g), H20(l), N2(g), S02(g), and HCl(aq). To convert to Btu/lb mol, multiply by 430.6.

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