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between the
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left) -

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reactions abound
in the geology
of Earth, in the
atmosphere and
oceans, and in a
vast array of
complicated
processes that
occur in all
living systems.
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reactions must be distinguished from physical changes.

Physical changes include changes of state, such as ice melting to water and water evaporating to vapour. If a physical change occurs, the

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physical

properties of a substance will change, but its chemical identity will remain the same.

chemical

reaction |

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500ml volumetric
flask with.

about 20ml of
distilled water.

2. Pipette 25ml
of hydrochloric
acid into. the
volumetric

flask. 3. By
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bottle, fill
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states that the
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a direction that tends to relieve the effect of an applied stress. Ex., an increase in temperature of a system alters the concentration relationship in the direction that tends to absorb heat.

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Chapter 9

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chapters cover many ways of making each type of molecule starting with simple aromatic and aliphatic compounds with one functional group and progressing to molecules with many functional groups.

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